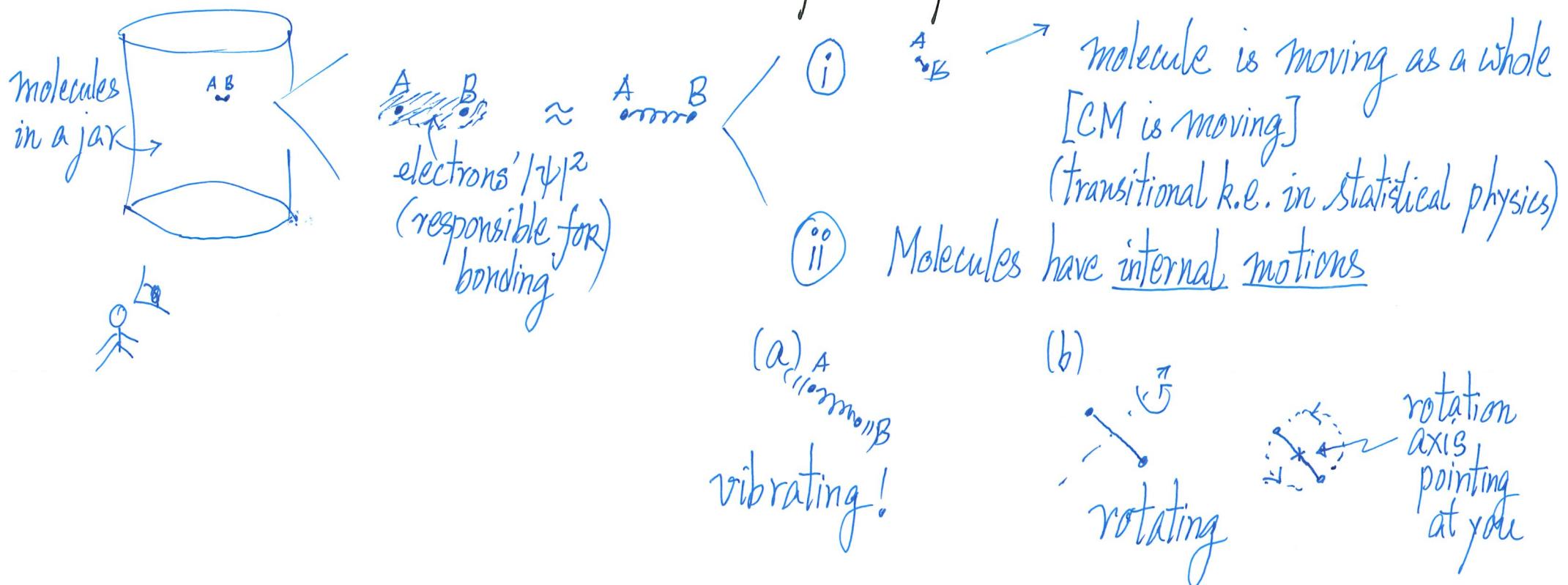
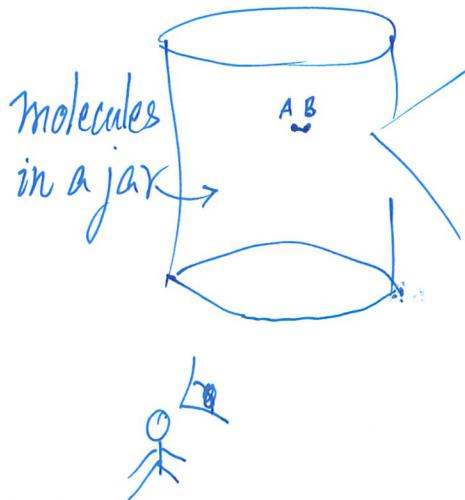


C. Quantum Mechanical Problems of Molecules are hard to solve

- Can't solve simplest molecular problem H_2
- Beauty of developing approximations

Diatom "AB" molecules ["simplest" problems]



In reality, everything happens at the same time!

Molecule flying in the room, vibrating and rotating as it flies!

Remark: This is the part that gives

$$U = \frac{3}{2} N k T$$

in statistical physics of (ideal) gas]

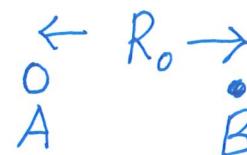
In QM: Don't worry about it!

Why? CM motion is free motion

Focus on relative motion

c.f.: Didn't worry about H-atom moving around

QM problems



"bond length"

Schrödinger Eq. for electrons

→ preferred separation R_0 ?

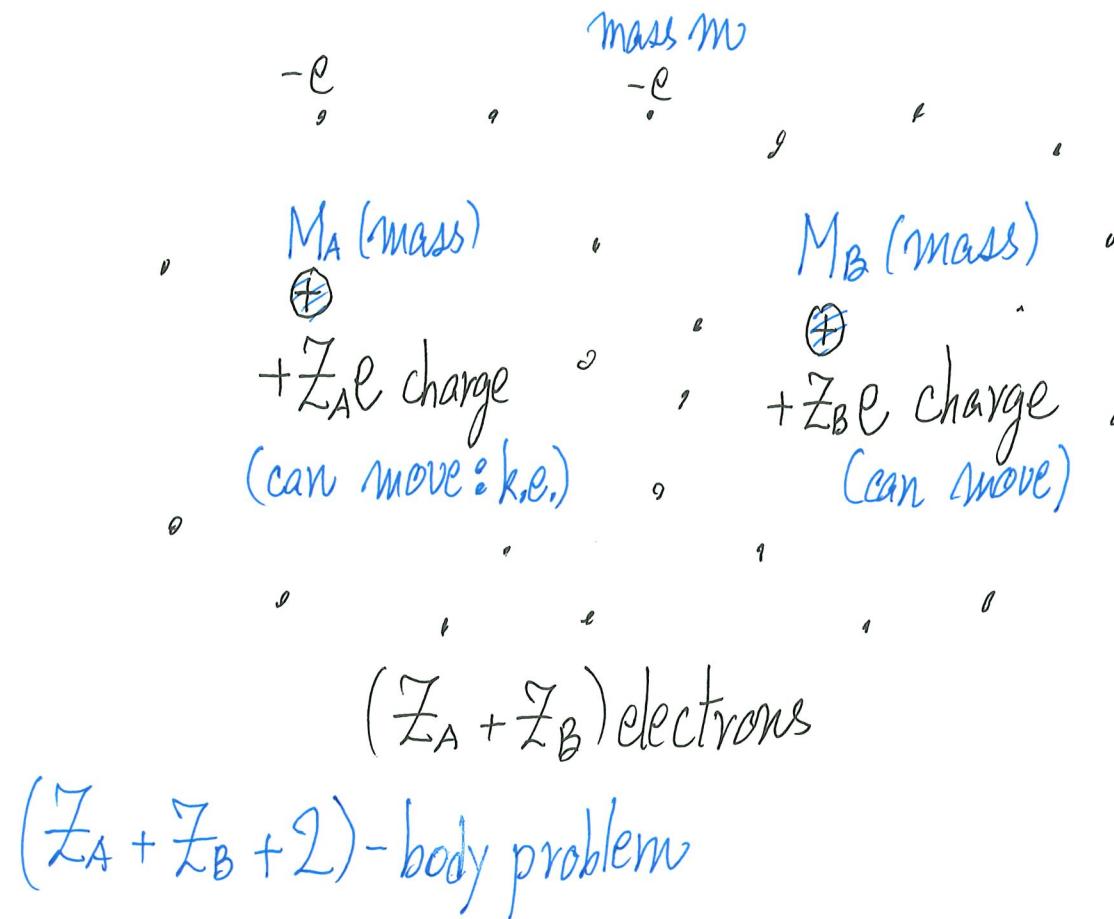
(rotation)

QM of oscillator



QM of (3D) rotor

Diatom "AB" Molecules



Hamiltonian includes:

- k.e. of nuclei
- k.e. of electrons
- Each electron sees A
- Each electron sees B
- Electron-electron repulsion
- Nucleus-nucleus repulsion

- We know the Hamiltonian and the governing Equation (TISE)
- $\hat{H}\bar{\Psi} = E\bar{\Psi}$ is hard to solve \Rightarrow Needs approximations to proceed

Full Hamiltonian of "AB" molecule

$$\begin{aligned}
 \hat{H} = & \left[-\frac{\hbar^2}{2M_A} \nabla_{\vec{R}_A}^2 - \frac{\hbar^2}{2M_B} \nabla_{\vec{R}_B}^2 \right] + \sum_{i=1}^{N=(Z_A+Z_B)} \left(-\frac{\hbar^2}{2m} \nabla_{\vec{r}_i}^2 \right) + \frac{Z_A Z_B e^2}{4\pi\epsilon_0 |\vec{R}_A - \vec{R}_B|} \\
 & \text{k.e. of nuclei A \& B} \quad \text{k.e. of } Z_A + Z_B = N \text{ electrons} \quad \text{p.e. (Coulomb repulsion) of nuclei} \\
 & - \sum_{i=1}^N \frac{Z_A e^2}{4\pi\epsilon_0 |\vec{r}_i - \vec{R}_A|} - \sum_{i=1}^N \frac{Z_B e^2}{4\pi\epsilon_0 |\vec{r}_i - \vec{R}_B|} + \sum_{i < j} \frac{e^2}{4\pi\epsilon_0 |\vec{r}_i - \vec{r}_j|} \\
 & \text{electrons see nucleus A (p.e.)} \quad \text{electrons see nucleus B (p.e.)} \quad \text{electron-electron Coulomb p.e.} \\
 & \text{(pairs of e's)}
 \end{aligned} \tag{1}$$

Z_A, Z_B = Atomic numbers (A & B) ; \vec{R}_A, \vec{R}_B = positions of nuclei A & B
 \vec{r}_i = position of i^{th} electron ; $(Z_A + Z_B + 2)$ -body problem ($(Z_A + Z_B)$ electrons + 2 nuclei)

Note: \vec{r}_i (electrons) and $\vec{R}_A (\vec{R}_B)$ appear together \Rightarrow Can't separate electron problem and nuclear motion problem

$$\hat{H} = \hat{H}\left(\overbrace{\hat{P}_A, \hat{R}_A; \hat{P}_B, \hat{R}_B}^{\text{nuclei}}, \overbrace{\hat{p}_1, \hat{r}_1; \hat{p}_2, \hat{r}_2; \dots; \hat{p}_N, \hat{r}_N}^{\text{electrons}}\right)$$

$$= \hat{H}\left(\hat{P}_A, \hat{R}_A; \hat{P}_B, \hat{R}_B; \{\hat{p}_i, \hat{r}_i\}\right) \quad (2)^+$$

Time-independent Schrödinger Equation

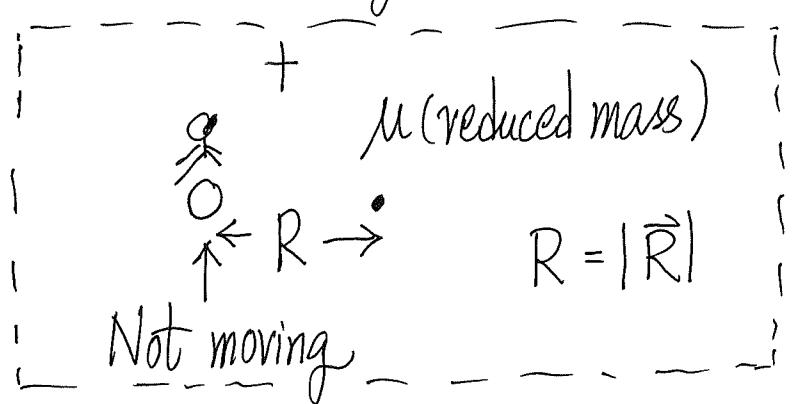
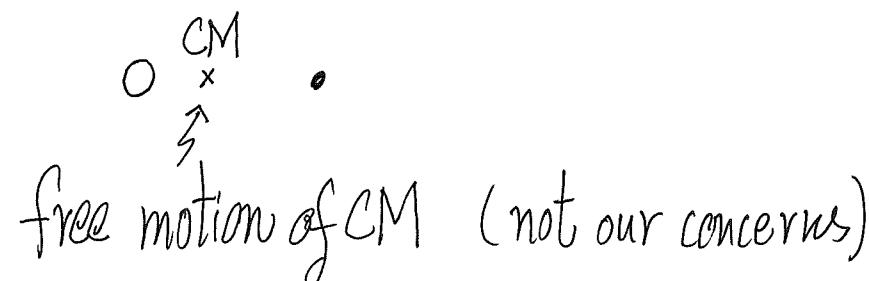
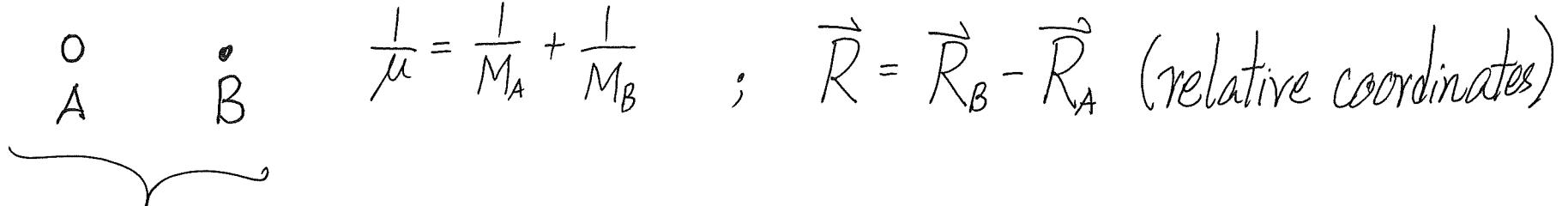
$$\hat{H}\left(\hat{P}_A, \hat{R}_A; \hat{P}_B, \hat{R}_B; \{\hat{p}_i, \hat{r}_i\}\right) \Psi_{\text{molecule}}(\vec{R}_A, \vec{R}_B; \vec{r}_1, \vec{r}_2, \dots, \vec{r}_N)$$

$$= E_{\text{total}}^{(\text{molecule})} \Psi_{\text{molecule}}(\vec{R}_A, \vec{R}_B; \vec{r}_1, \vec{r}_2, \dots, \vec{r}_N)$$

to solve for (many) allowed energies of a molecule $E_{\text{total}}^{(\text{molecule})}$ and corresponding molecular state Ψ_{molecule} .

⁺Atomic problems are "easier" as $\hat{H}_{\text{atom}}(\{\hat{p}_i, \hat{r}_i\})$ only, with the nucleus at the origin.

- Done! We have TISE (Eq.(3))! All physics has been used.
- Done? Eq.(3) can't be solved for any diatomic ("AB") molecule!
- Even single out the CM (center-of-mass) motion won't help!



one less (CM) coordinate
 [still $\underbrace{N(\text{electrons}) + 1(\text{relative, nucleus})}_{(N+1)-\text{body}}$ problem]

Retaining relative (nuclear) motion and $N(Z_A + Z_B)$ electrons

$$\hat{H}_{\text{total}} = \frac{-\hbar^2}{2\mu} \nabla_{\vec{R}}^2 + \sum_{i=1}^N \left(\frac{-\hbar^2}{2m} \nabla_{\vec{r}_i}^2 \right) + \frac{Z_A Z_B e^2}{4\pi\epsilon_0 R} - \sum_{i=1}^N \left(\frac{Z_A e^2}{4\pi\epsilon_0 |\vec{r}_i - \vec{R}_A|} + \frac{Z_B e^2}{4\pi\epsilon_0 |\vec{r}_i - \vec{R}_B|} \right) + \sum_{i < j} \frac{e^2}{4\pi\epsilon_0 |\vec{r}_i - \vec{r}_j|} \quad (4)$$

Now, TISE becomes

$$\boxed{\hat{H}_{\text{total}} \Psi_{\text{molecule}}(\vec{R}; \vec{r}_1, \dots, \vec{r}_N) = E_{\text{total}}^{(\text{molecule})} \Psi_{\text{molecule}}(\vec{R}; \vec{r}_1, \dots, \vec{r}_N)} \quad (5)$$

This is the
Full Diatomic
Molecule
problem

Difficulties $\frac{-\hbar^2}{2\mu} \nabla_{\vec{R}}^2 \neq \frac{\hat{P}^2}{2\mu}$; $\sim \frac{1}{|\vec{r}_i - \vec{R}|} = \frac{1}{|\vec{r}_i - \hat{\vec{R}}|}$
momentum of nucleus coupled!
operators!

CANNOT separate \hat{H}_{total} into terms with nuclear operators only and terms with electron operators only

Implications

function of \vec{R} (nuclear) AND $\{\vec{r}_i\}$ (electrons)

- Ψ_{molecule} is formally $\Psi_{\text{molecule}}(\vec{R}; \vec{r}_1, \dots, \vec{r}_N)$
- Formally⁺ $\Psi_{\text{molecule}}(\vec{R}, \vec{r}_1, \dots, \vec{r}_N) \neq \underbrace{\psi_N(\vec{R})}_{\substack{\uparrow \\ (\text{Nucleus})}} \cdot \underbrace{\psi_{\text{el}}(\vec{r}_1, \dots, \vec{r}_N)}_{\substack{\text{electrons' coordinates only}}} \quad (6)$

[⁺ Recall: When separation of variables works]

- But it will be convenient (thus desirable) to separate nuclear and electronic parts

Key idea → Way forward? Make the approximation that

$$\Psi_{\text{molecule}}(\vec{R}; \vec{r}_1, \dots, \vec{r}_N) \approx \underbrace{\psi_N(\vec{R})}_{\substack{\text{vibration \& rotation} \\ (\text{nuclear motion})}} \cdot \underbrace{\psi_{\text{el}}(\vec{r}_1, \dots, \vec{r}_N)}_{\substack{\text{electrons' coordinates only} \\ \text{nuclei NOT moving (fixed } R \text{)}}} \quad (7)$$